







## Withdrawal from 2007-02-01 - SCAN-test methods of chemical character

Manganese content	SCAN-CM 14:05 (withdrawn)	ISO 1830:2005 (replacing SCAN)
Applicable to	Pulp and paper with ash content $\leq$ 3% acc to ISO 1762.	Paper, board and pulps (the acid-soluble part of the ignition residue acc to ISO 1762).
Principle	Ignition at 525 °C, dissolution in HNO <sub>3</sub> , photometric end-determination. (AAS end-determination in SCAN-CM 38).	Ignition at 525 °C, dissolution in HCl. End-determination: flame AAS or ICP (inductively coupled plasma spectroscopy)
Procedure		
Ignition	525 ± 25 °C (ISO 1762)	525 ± 25 °C (ISO 1762)
Dissolution	In 1,5 mol/l HNO <sub>3</sub>	In 6 mol/l HCl
Oxidation	Periodate	-
End-determination	Spectoophotometer at 525 nm	Flame AAS, the 279,5 nm line ICP, the 257,6 nm line
Duplicate determinations	Yes	Yes
Report	In mg/kg >1 mg/kg: 2 significant figures <1 mg/kg: 1 significant figure	In mg/kg with 2 significant figures

Total chlorine	SCAN-CM 51:91 SCAN-P 68:94 (withdrawn)	ISO 11480:1997 (replacing SCAN)
Applicable to	Pulp, paper and board	Pulp, paper and board. Lower limit of determination 20 mg/kg
Principle	Procedure A: Schöniger combustion + IC or Schöniger combustion + titration	No Schöniger procedure
	Procedure B (AOX apparatus): Combustion in a heated quartz tube + micro-coulometric titration	Combustion in a heated quartz tube + micro-coulometric titration
Procedure A	Samples with results >20 mg/kg (IC) Samples with results >200 mg/kg (titr)	
Duplicate determinations	Yes	No Schöniger procedure
Sample size	250 – 500 mg	
Combustion	In O <sub>2</sub> in Schöniger flask	
Schöniger flask, volume	750 – 1000 ml	
Ion chomatography	Fractionating column, chloride, conductivity detector	
Potentiometric titration with AgNO <sub>3</sub> solution	Silver electrode + mercury sulphate electrode or double junction calomel electrode	
Procedure B (AOX apparatus)		AOX apparatus
Duplicate determinations	Yes	Yes
Sample size	10 – 60 mg	10 – 60 mg
Combustion	In O <sub>2</sub> in a quartz tube	In O <sub>2</sub> in a quartz tube

Absorption	HCl is absorbed in electrolyte solution	HCl is absorbed in electrolyte solution
Micro-coulometric	Yes	Yes
titration		
Report	In mg/kg with 2 significant figures	In mg/kg with 2 significant figures
	(procedure A and B)	

Organic chlorine	SCAN-CM 52:91	ISO 11480:1997
	SCAN-P 69:94	(replacing SCAN)
	(withdrawn)	
Applicable to	Pulp, paper and board with OCl content	Pulp, paper and board. Lower limit of
	> 20 mg/kg	determination 20 mg/kg.
Principle	Procedure A:	
	Schöniger combustion + IC	No Schöniger procedure
	<i>Procedure B</i> (AOX apparatus):	AOX procedure
	Combustion in heated tube + micro-	Combustion in heated tube + micro-
	coulometric titration	coulometric titration
Procedure A	OCL content: 100 - 25000 mg/kg	
Duplicate determinations	Yes	
Sample size	200 – 500 mg	
Washing	With NaNO <sub>3</sub>	
Filtration	Yes	No Schöniger procedure
Combustion	In O <sub>2</sub> in Schöniger flask	
Schöniger flask, volume	750 – 1000 ml	
Ion chomatography	Fractionating column, chloride,	
	conductivity detector	
Procedure B	OCL content: 10 - 25000 mg/kg	
Duplicate determinations	Yes	Yes
Sample size	10 – 60 mg	10 - 60  mg
Washing	With NaNO <sub>3</sub>	With NaNO <sub>3</sub> and activated Carbon
Filtration	Yes	Yes
Combustion	In O <sub>2</sub> in a quartz tube	In O <sub>2</sub> in a quartz tube
Absorption	HCl is absorbed in electrolyte solution	HCl is absorbed in electrolyte solution
Micro-coulometric	Yes	Yes
titration		
Report	In mg/kg with 2 significant figures	In mg/kg with 2 significant figures
	(procedure A and B)	

pH of aqueous extract	SCAN-P 14:65 (withdrawn)	ISO 6588-1:2005 Cold extraction (ISO 6588-2:2005 Hot extraction) (replacing SCAN)
Applicable to	Paper	Paper board and pulps
Principle	Soaking in cold distilled water. The pH is measured.	Soaking in cold distilled water. Filtration and addition of a salt solution. The pH is measured.
Procedure		·
Sample size	1,0 g	$2.0 \pm 0.1$ g
Temperature	25 ± 5 °C	20 - 25 °C
Distilled water volume	20 ml + 50 ml	100 ml
Stirring time	30 s	-
Soaking time	1 h	1 h
Filtration	No	Yes + addition of a salt solution
pH measurement	Glass electrode	Glass and calomel electrodes or a

		combined ectrode
Duplicate determinations	Yes	Yes
Report	To the nearest 0,1 pH unit	To the nearest 0,1 pH unit